## **Molecule Building Simulation**

Name	Block

- 1) Find the website for pHET simulations.
- 2) Open the program entitled "Molecule Shapes" under Chemistry Programs.
- 3) Draw the Lewis diagram of the following molecules before building them on the simulation.
- 4) Build the molecules on the simulation (make sure to include lone pairs of electrons on the central atom)
- 5) Check the boxes for "Molecule Geometry" and "Show bond angles" and fill in the following table. Give a brief sketch of 3-D drawing.
- 6) Predict whether the molecule is polar or not. (polar = uneven distribution of charges in a molecule.)

Molecule	Lewis Diagram	Bond°	Shape Name	3-D Drawing	Polar?
H <sub>2</sub>					N
CIF <sub>3</sub>					Y
XeF <sub>4</sub>	-		•	•	$\sim$
H <sub>2</sub> O					Y
SF <sub>6</sub>	4 <sup>62</sup>				✓
HCN					Y
CH <sub>2</sub> O					Y